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## Preliminary phytochemical screening, isolation and structural elucidation of chloroform leaf extracts of *Maesa lanceolata*

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### ABSTRACT

The aim of this study was to screen the phytochemical constituents, isolate and elucidate the structure of chloroform extracts from the leaves of *Maesa lanceolata*. The qualitative phytochemical analysis of the leaf extract of *Maesa lanceolata* was done following standard procedures and the tests revealed the presence of cardiac glycosides and saponins in the crude extracts. The chloroform extract of the leaves of *M. lanceolata* led to the isolation of one compound, C1. Structure determination was accomplished by means of spectroscopic methods (IR,  $^{13}\text{C}$  and  $^1\text{H}$  NMR). According to the spectral data (IR,  $^1\text{H}$ NMR,  $^{13}\text{C}$ NMR and DEPT), probably Myrsenene was identified which is reported for the first time from *M. lanceolata*.

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**Capsule Summary:** Phytochemical constituent's analysis, isolation and structure elucidation of chloroform extracts from the leaves of *Maesa lanceolata* were investigated.

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### INTRODUCTION

*Maesa lanceolata* (*Myrsinaceae*) is a well-known plant and widely distributed in many parts of Africa especially in Ethiopia (Leulet al., 2010; Dereket al., 2009). The leaves of the plant traditionally used for the treatment of helminthes and bacterial infections in most rural areas of the country since it is rich in biologically active constituents (Takele et al., 2014; Focho et al., 2009). In the previous studies, researchers reported the isolation of hydroxylated-1, 4-benzoquinone derivatives from various parts of plant. Qualitative phytochemical screenings of the different solvent extracts of the plant undertaken by different investigators have revealed the isolation and identification of various acylated triterpene

saponins based on oleanane skeleton. Lately, flavonol glycosides and alkaloids have been reported from the plant leaves (Lawrence et al., 2011). In the present study, preliminary phytochemical analysis, isolation and structural elucidation of chloroform extract of the plant leaves has been done.

### MATERIAL AND METHODS

#### Sample collection and preparation

The fresh leaves of *Maesa lanceolata* plant were collected from Awi Zone, Guangua Wereda at Elala forest which is about 220 km away from Bahir Dar City during December 2014 (Fig 1). The collected plant materials were properly



**Fig 1:** The photograph of *Maesa lanceolata* Leaf

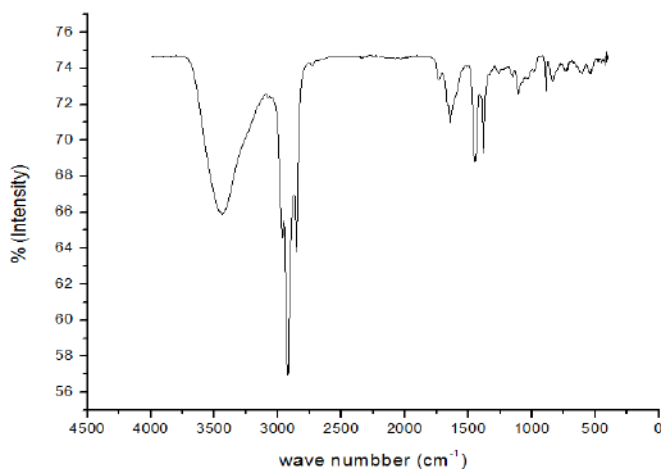
washed with tap water, air dried for two weeks under shade. The dry leaves were ground to obtain a fine powder.

#### Extraction and isolation of *Maesa lanceolata* leaves

150 g fresh and uncrushed leaves of *Maesa lanceolata* were successively extracted with 250 mL of chloroform at room temperature. The extracts were freed from the solvent under reduced pressure to give crude extract. The combined extract was subjected to column chromatography. As indicated in Table 1, the column was eluted first with hexane then with hexane: EtOAc (5:1, 2:1, 1:1, 1:2 and 1:5) and EtOAc to pure and get rid of any EtOAc retained in the packed column. This gave different fractions and TLC of each fraction was examined using different solvent systems to give four major fractions (I-V). The first two fractions contained oily substances and chlorophyll and were discarded. Fractions III to V were further purified using column chromatography and refractionated using different solvent systems. Purification by preparative TLC was done for each fraction to give pure

**Table 1:** Solvent system (*n*-hexane/EtOAc) used to separate compound C<sub>1</sub>

Solvent system (ratio)	Total amount taken (mL)
100 %	30
5:1	30
2:1	30
1:1	30
1:2	30
1:5	30
100%	30



**Fig 2:** FTIR spectra of compound C<sub>1</sub>

compound, C<sub>1</sub>. The correct structural of pure component (C<sub>1</sub>) was elucidated by NMR (<sup>1</sup>H-NMR, <sup>13</sup>C-NMR, DEPT) and IR techniques (Ayoola et al., 2008; Zohra et al., 2012; Tiwari et al., 2011, Carmen-María et al., 2014; Abdulet al., 2015; Harami et al., 2016).

#### Phytochemical screening of the extract

The preliminary phytochemical screening of the extracts of *Maesa lanceolata* leaves have been done following standard procedures. The leaf extract was subjected to qualitative chemical investigation using specific reagents for each group of constituents (Margeretha et al., 2012; Beatrice et al., 2002). Compound C<sub>1</sub>. R<sub>f</sub> value of 0.75cm (CHCl<sub>3</sub>); light yellow, IR (KBr)  $\nu_{max}$  3300, 3000, 2900, 1600, 1460, 1380, 1250 and 720 cm<sup>-1</sup>; <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD)  $\delta$  (ppm)  $\delta$  0.80, 1.05, 1.30, 1.45, 1.65, 1.70, 1.95, 2.1, 5.15 and 4.55 – 4.85. <sup>13</sup>CNMR and DEPT spectrum of compound C<sub>1</sub> is summarized in Table 2.

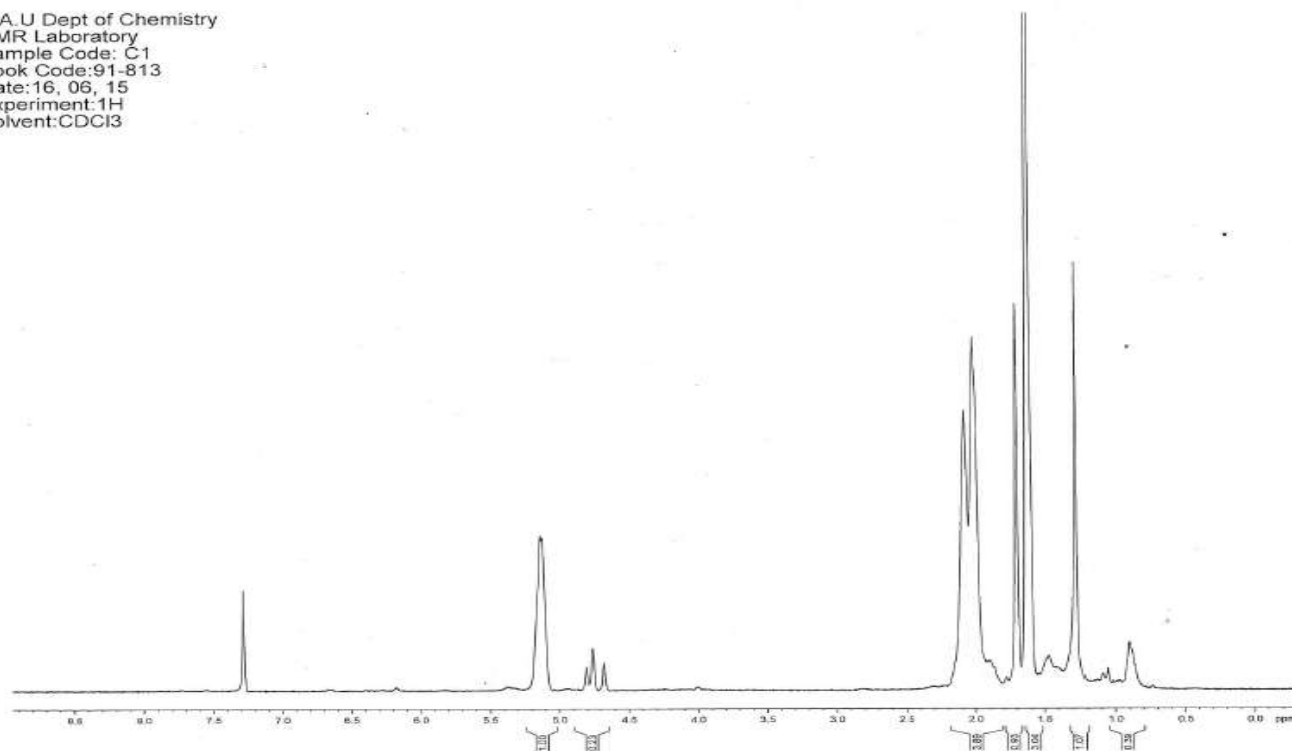
## RESULTS AND DISCUSSION

#### Phytochemical analysis of the extract

The results of phytochemical screening test performed on crude leaf extracts of *Maesa lanceolata* plant are summarized in Table 3 below. Phytochemical analysis of chloroform crude extracts of *Maesa lanceolata* leaf extract revealed the presence of cardiac glycosides and saponins. The presence of these substances in the investigated plant accounts for its usefulness as medicinal plant. This information obtained is used to facilitate quantitative estimation and qualitative separation of constituents from the leaves. In addition to the phytochemical screening of the plant extract, we have checked the anthelmintic activities and the extract showed the prominent activity towards aquatic leech; *Lymnatis nilotica* (Wondu and Alemayehu, 2016).

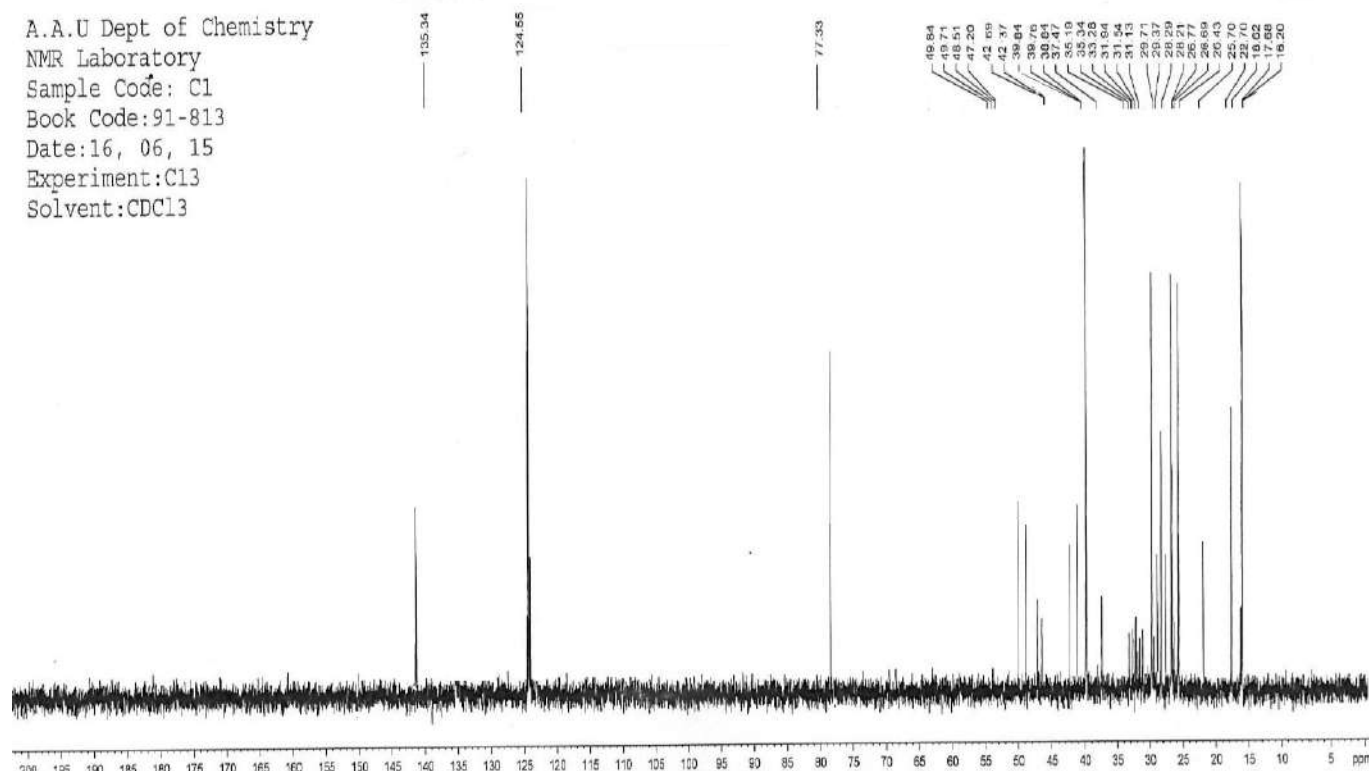
#### FTIR analysis

A.A.U Dept of Chemistry  
NMR Laboratory  
Sample Code: C1  
Book Code:91-813  
Date:16, 06, 15  
Experiment:1H  
Solvent:CDCl3



**Fig 3:**  $^1\text{H}$ -NMR spectra of  $\text{C}_1$

A.A.U Dept of Chemistry  
NMR Laboratory  
Sample Code: C1  
Book Code:91-813  
Date:16, 06, 15  
Experiment:C13  
Solvent:CDCl3



**Fig 4:**  $^{13}\text{C}$ -NMR spectra of compound  $\text{C}_1$



**Table 2:**  $^{13}\text{C}$  NMR and DEPT spectrum of compound C<sub>1</sub>

Carbon Number	Multiplicity (DEPT)	$^{13}\text{C}$ -NMR	Remark
1	CH <sub>2</sub>	35.9	Methylene
2	CH <sub>2</sub>	29.71	Methylene
3	CH	77.33	Methine
4	C	37.47	Quaternary Carbon
5	CH	47.20	Methine
6	CH <sub>2</sub>	28.29	Methylene
7	CH <sub>2</sub>	49.5	Methylene
8	C	39.76	Quaternary Carbon
9	C	135.34	Quaternary Carbon
10	C	39.84	Quaternary Carbon
11	CH	124.55	Methine
12	CH <sub>2</sub>	28.21	Methylene
13	CH	49.71	Methine
14	C	42.35	Quaternary Carbon
15	CH <sub>2</sub>	31.94	Methylene
16	CH <sub>2</sub>	33.25	Methylene
17	C	38.84	Quaternary Carbon
18	CH	48.51	Methine
19	CH <sub>2</sub>	42.56	Methylene
20	C	29.37	Quaternary Carbon
21	CH <sub>2</sub>	31.54	Methylene
22	CH <sub>2</sub>	35.54	Methylene
23	CH <sub>3</sub>	26.43	Methyl
24	CH <sub>3</sub>	25.70	Methyl
25	CH <sub>3</sub>	18.62	Methyl
26	CH <sub>3</sub>	22.70	Methyl
27	CH <sub>3</sub>	17.68	Methyl
28	CH <sub>3</sub>	26.77	Methyl
29	CH <sub>3</sub>	31.13	Methyl
30	CH <sub>3</sub>	26.69	Methyl

The FTIR analysis of the sample (C<sub>1</sub>) revealed the presence of polycyclic compound with alcohol functional group (Fig 2). Accordingly from the FTIR spectrum, the prominent absorption band at 3300 cm<sup>-1</sup> indicates the O-H stretch. The absorption band at 3000 cm<sup>-1</sup> indicates sp<sup>2</sup> C-H stretch with weak intensity and the absorption band at 2900 cm<sup>-1</sup> reveals the presence of sp<sup>3</sup> C-H stretch with weak intensity. The FTIR spectrum also showed the presence of olefinic carbon (C=C) at 1600 cm<sup>-1</sup> with medium intensity and the presence of -CH<sub>2</sub> out of plane bending with weak intensity at 1460 cm<sup>-1</sup>. Furthermore the absorption band at 1380 cm<sup>-1</sup> indicates -CH<sub>3</sub> bending with medium intensity and there is weak absorption band of C-O stretch at 1250 cm<sup>-1</sup>. The weak absorption band of -CH<sub>2</sub> rocking was also observed at 720 cm<sup>-1</sup>.

$^1\text{H}$ -NMR spectrum of the compound showed eight singlet signals due to the presence methyl groups at  $\delta$  0.80, 1.05, 1.30, 1.45, 1.65, 1.70, 1.95 and 2.1, which are due to the hydrogen atoms attached to C-29, C-30, C-28, C-27, C-25, C-26, C-24 and C-23, respectively. One proton signal at  $\delta$  5.15 is due to methine proton at C-11. A triplet proton signal at  $\delta$  4.55 – 4.85 is due to the presence of protons at C-12. The  $^1\text{H}$ -NMR data, therefore, clearly indicates compound C<sub>1</sub> is triterpene.

#### $^{13}\text{C}$ NMR and DEPT spectrum

The proton decoupled  $^{13}\text{C}$ -NMR spectrum (both 1D  $^{13}\text{C}$ -NMR and DEPT) showed the signals of 30 carbon atoms (Table 3) of eight methyl (C-23 to C-30), ten methylene (C-1, C-2, C-4, C-7, C-12, C-15, C-16, C-19, C-19, C-21 and C-22), five methine (C-3, C-5, C-11, C-13 and C-18) and seven quaternary carbons

(C-6, C-8, C-9, C-10, C-14, C-17 and C-20). In DEPT spectrum, the data are collected in such way that the resulting signal is either positive (CH & CH<sub>3</sub>) or negative (CH<sub>2</sub>) depending on the number of protons attached. Accordingly, ten signals are pointing down (negative) indicating that there are ten methylene groups in the C<sub>1</sub> and eight signals pointing upward (positive) indicating eight carbons attached with either one or three hydrogens. Based on the information obtained, the most likely structure of the isolated compound (C<sub>1</sub>) is found to *Myrsenene* (Fig. 6).

**Table 3:** Preliminary phytochemical analysis of chloroform extracts of *Maesa lanceolata*

S.No	Phytochemicals	Result
1.	Tannins	-
2.	Cardiac Glycosides	+
3.	Alkaloids (Wondu and Alemayehu, 2016)	+
4.	Flavonoids(Wondu and Alemayehu, 2016)	+
5.	Steroids	-
6.	Anthraquinones	-
7.	Saponins	+
8.	Proteins	-

## CONCLUSIONS

Qualitative preliminary phytochemical investigation of chloroform extracts of the leaf of *Maesa lanceolata* revealed that the presence of saponins and cardiac glycosides. Upon isolation and structural elucidation of the extract, one compound was found with the name of *Myrsenene*.

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